



THE UNIVERSITY
of ADELAIDE

VOLKER HESSEL

THE CHEMICAL ENGINEERING CHALLENGES OF MINING ASTEROIDS

AUSIMM, ADELAIDE, 15.08.19

volker.hessel@adelaide.edu.au

CONTENTS

- ❑ Need of interdisciplinarity and technology disruption
- ❑ Historic learning: need for supply chains and hubs (business case)
- ❑ Micro-flow and space = Space-labs
- ❑ Asteroid mine atlas
- ❑ Continuous-flow extraction: coiled flow inverter for Co/Ni
- ❑ Disruptive opportunities for asteroid mining economics: high-c, high-T, ILs
- ❑ Space-mimicked microfluidics
- ❑ Holistic picture – orbital economics & deep space communication

Hung, Tele-communication



Sanaz, Geology



Mahdieh, IL extraction



Nam, Life cycle manag.



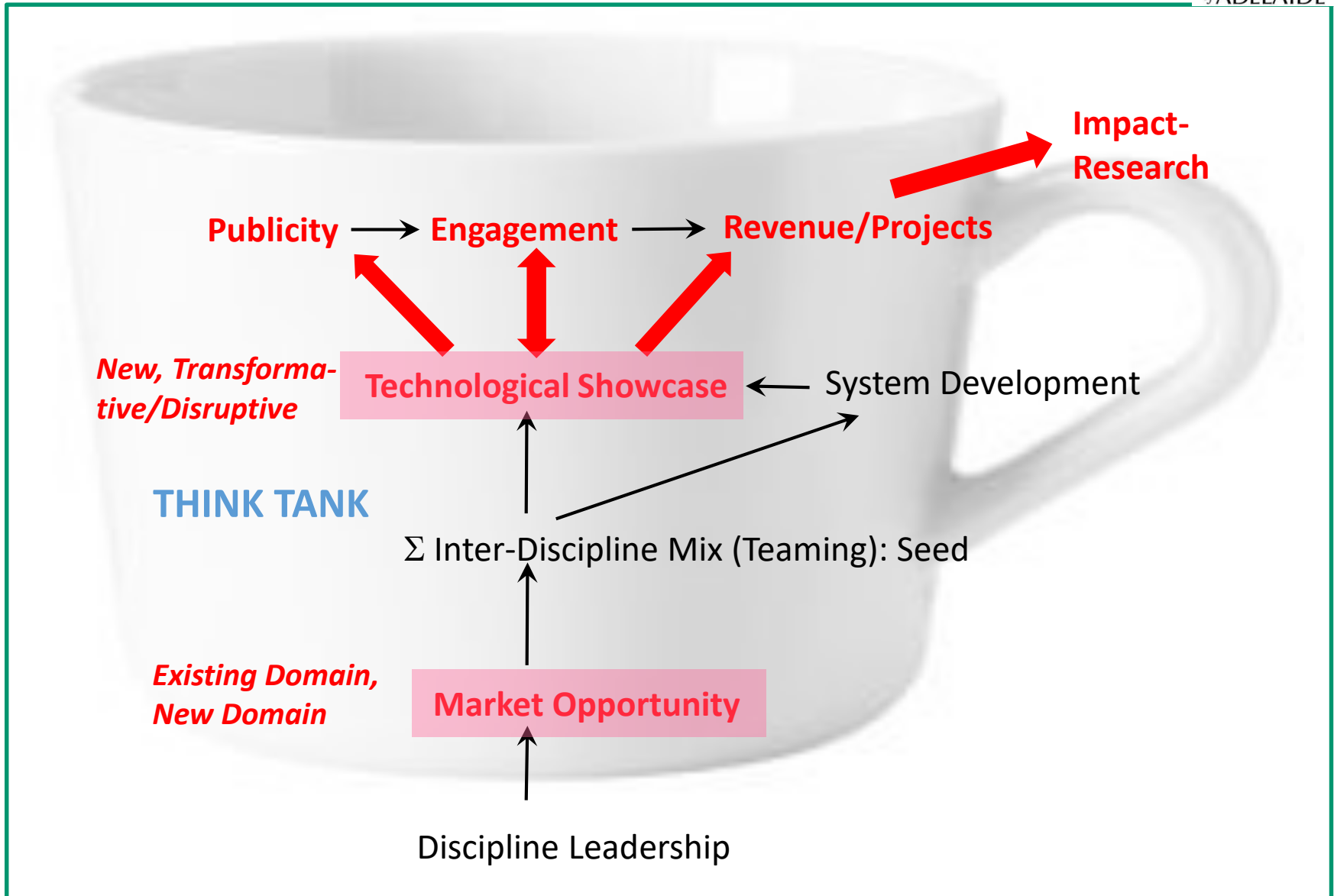
Volker, Chem-Eng disruption



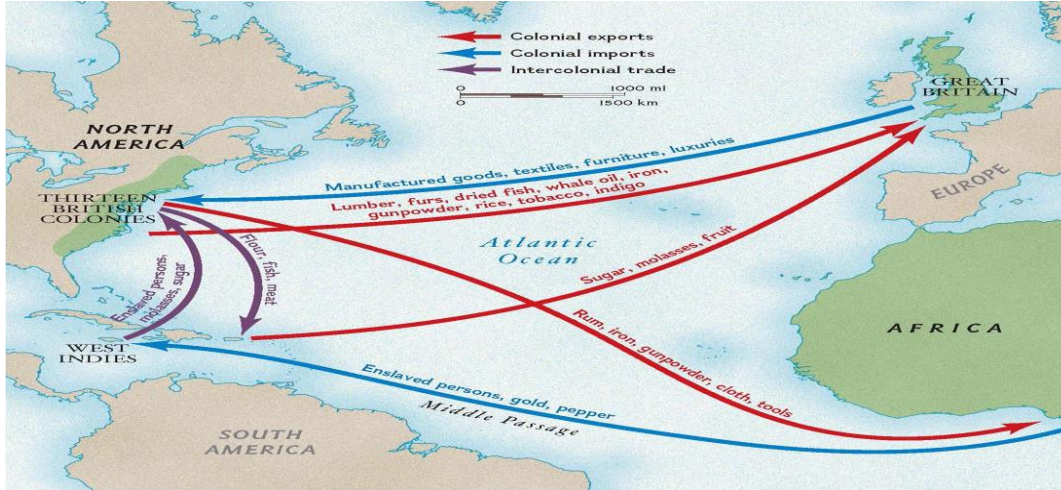
Wendy & Marko, Results fishing



THEMES – PROACTIVE THINK-TANKS



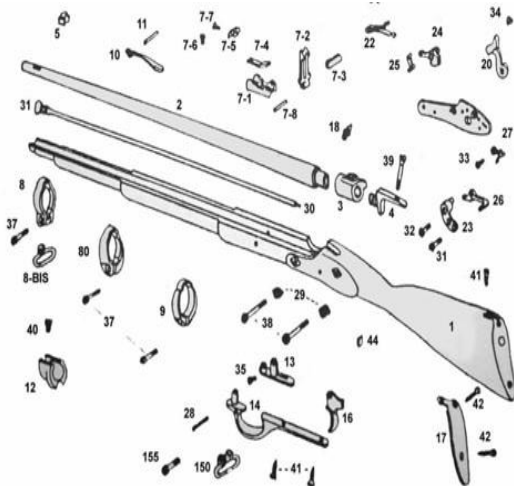
DISRUPTIVE TECHNOLOGY & INDUSTRIAL TRANSFORMATION OF THE AMERICAS AFTER 1492



Export through top sales product:
21 million beaver hats (1700-1770)

=> Export 1750: Germany: 16,500,
Spain: 110,000, Portugal: 175,000

Invention: Modularisation



Eli Whitney, 1801

Interchangeable parts

Transformed America
from **an artisan-based
nation to an assembly
line style producer**

- **New England** was more far and unknown
400 years than asteroids today
- **Supply chains** and **hubs**
- **New, unknown resources** (eg tobacco &
beaver hat): **disruptive technologies**
- **(Semi-) finished goods & innovation**
- **Transformative business window:
innovations**

CONTINUOUS BECOMES NEW STANDARD



in-Pharma
Technologist.com

FREE
Breaking |



FDA calls on manufacturers to begin switch from batch to continuous production

D. J. C. Constable, C. Jimenez-Gonzalez, R. K. Henderson, *Org. Process Res. Dev.* 11 (2007) 133-137

... ALSO IN SPACE ... NEW STANDARD



SpaceTango
(with Zaiput)

Flow Chemistry in Spaaaaaaace!

CSIRO

Flow Chemistry in Space—A Unique Opportunity to Perform Extraterrestrial Research

J. Flow Chem. 2017, 7(3–4), 151; Nat. Rev. Chem. 2017, 1, 0055

DECEMBER 1, 2015

Getting into the flow on the International Space Station

by Mike Giannone, NASA

www.spaceflow.org

American Chemical Society

The Space Summit
2019
Coronado Island/San Diego
14-15 October

Organs-on-Chips in Space

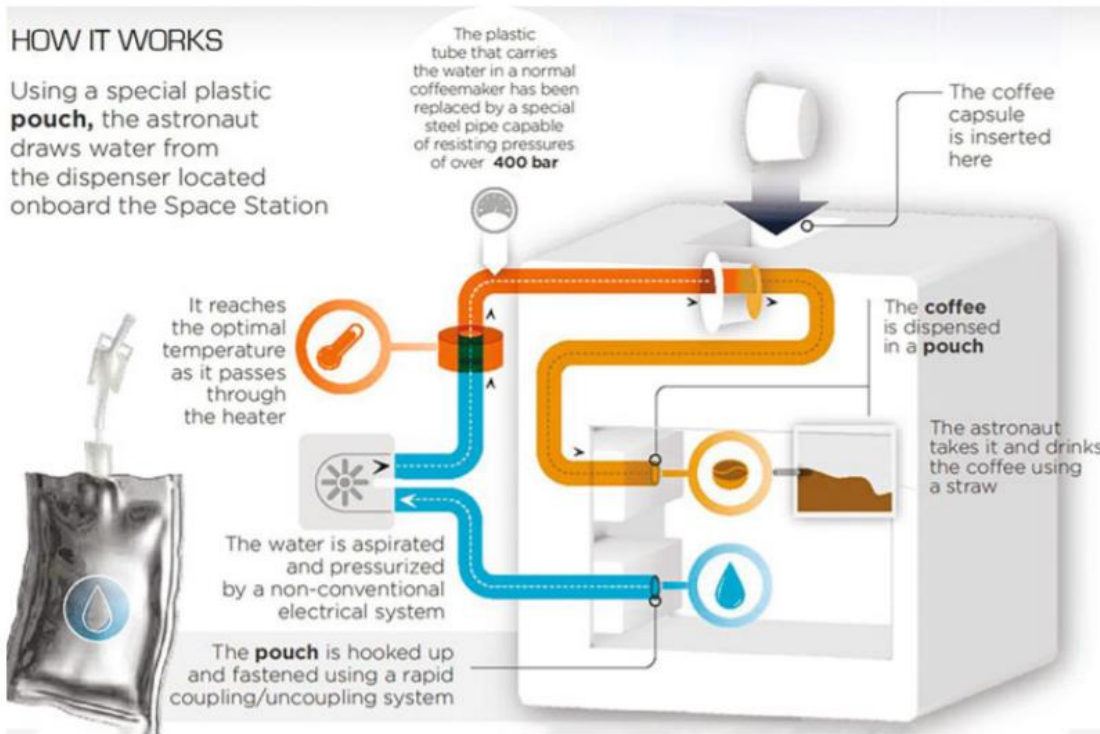
3D-Organoids in Space

PHARMA GOES SPACE

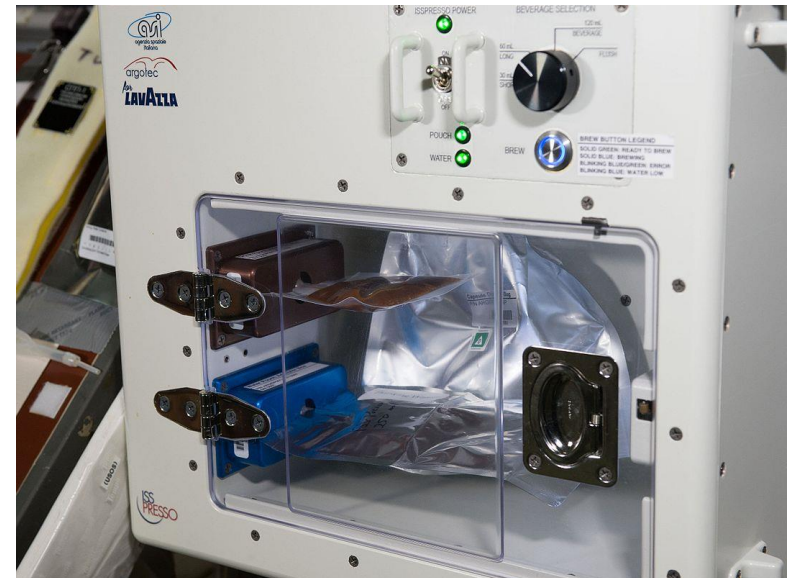
Pharma giants Merck, Procter & Gamble, Amgen and Eli Lilly are on International Space Station

SpacePharma and Space Tango offer 'space pharma laboratories'

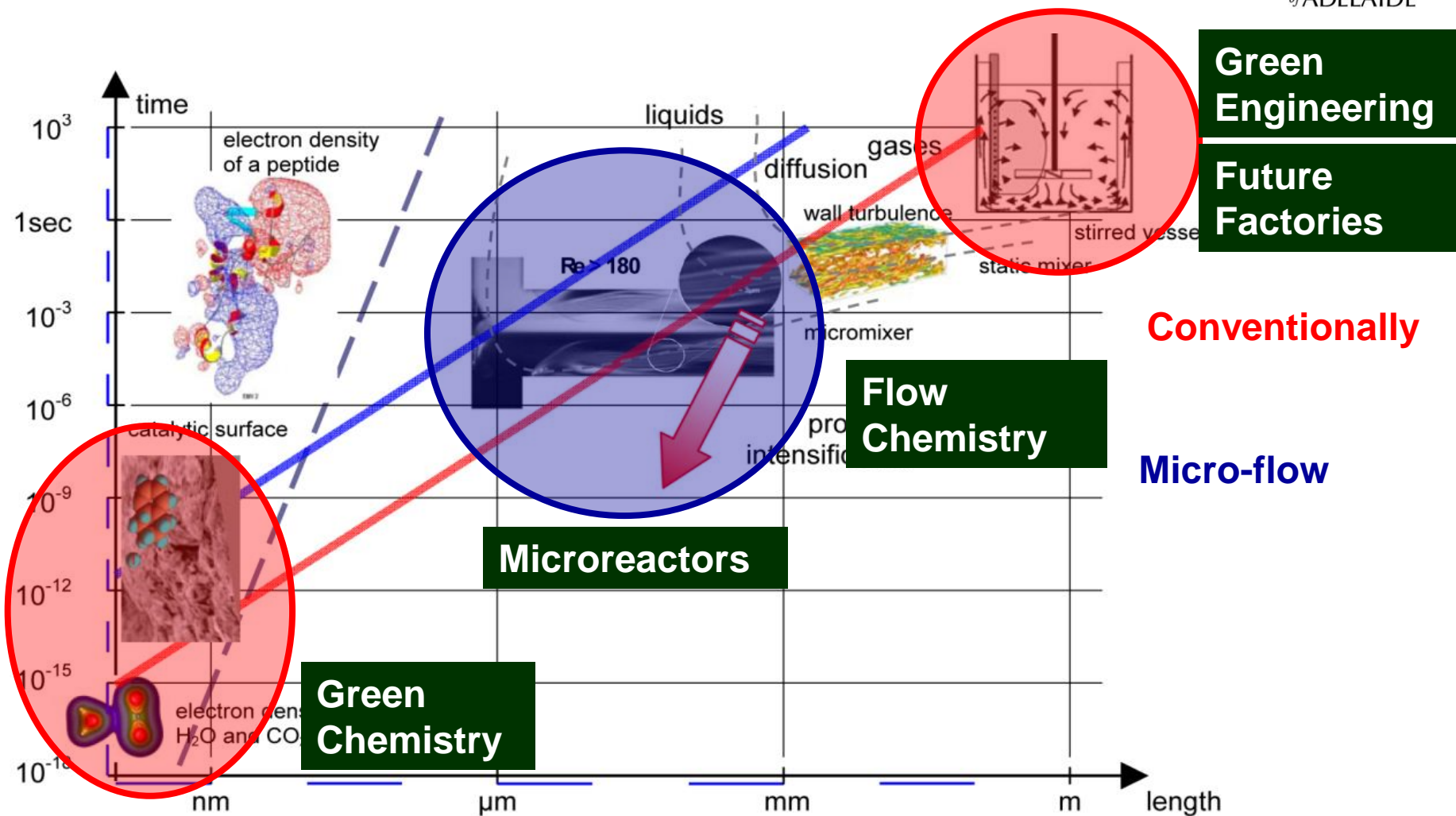
More rodents left the Earth's orbit than any other mammal



ISSpresso



LENGTH AND TIME: 'MULTI-SCALES'

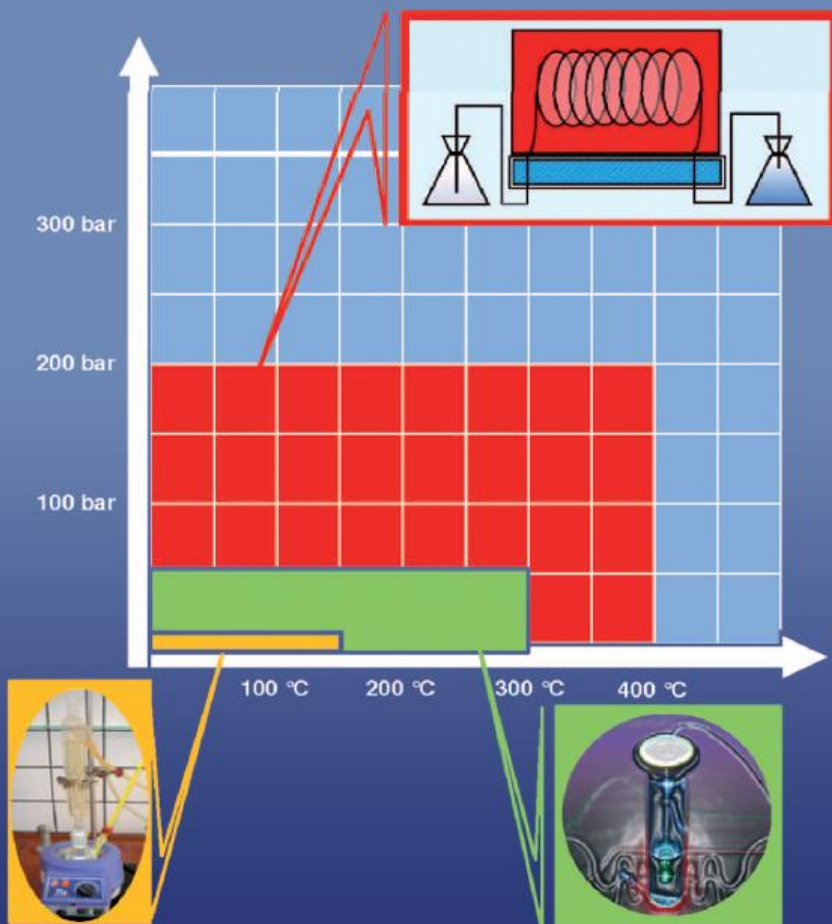


DISRUPTIVE ENGINEERING ON EARTH ... & IN SPACE

- Process intensification = processing at first principles**
- Novel Process Windows = Novel Business Windows**
- Lower costs and better environmental footprint**
- Automation = opportunity for artificial intelligence**
- Modularisation = LEGO for space challenges**
- Compactness / Light-weight = low payload**
- Zero gravity processing feasible = capillary forces dominate**
- Vacuum processing feasible = enclosed process chamber;
no headspace**

NOVEL PROCESS WINDOWS – HIGH-p,T: SUPERHEATED PROCESSING

High T,p flow processing



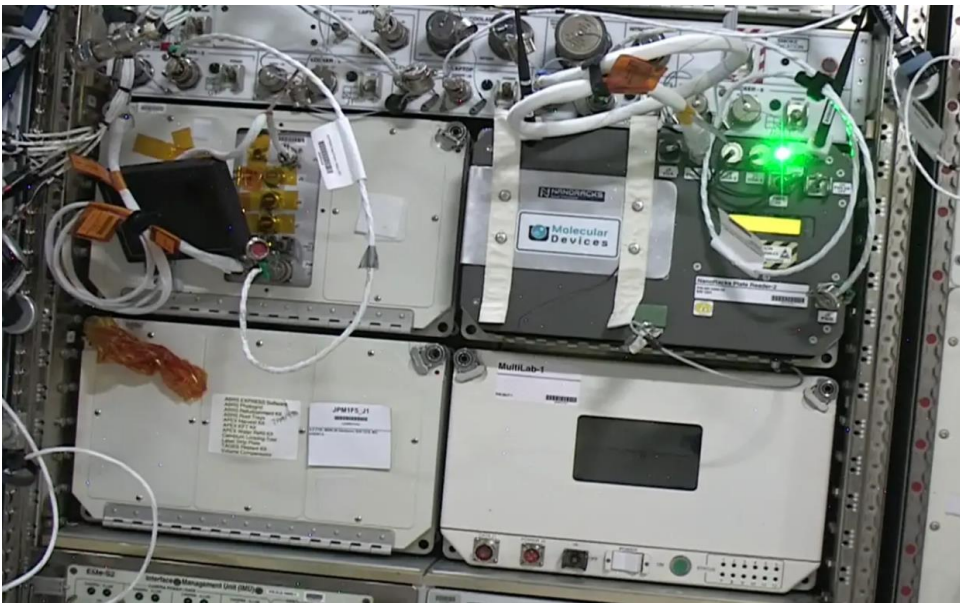
Microreactors expand the p, T process window and more (e.g., concerning c). Microwave did similar, but is difficult to scale-up

NOVEL PROCESS WINDOWS

V. Hessel, N. Kockmann, D. Kralisch,
T. Noel, Q. Wang, *ChemSusChem*
65 (2013) 746-789

T. Razzaq, C. O. Kappe, *Chem.
Asian J.* **5**, 6, 1274-128

SPACE MINI-LABS

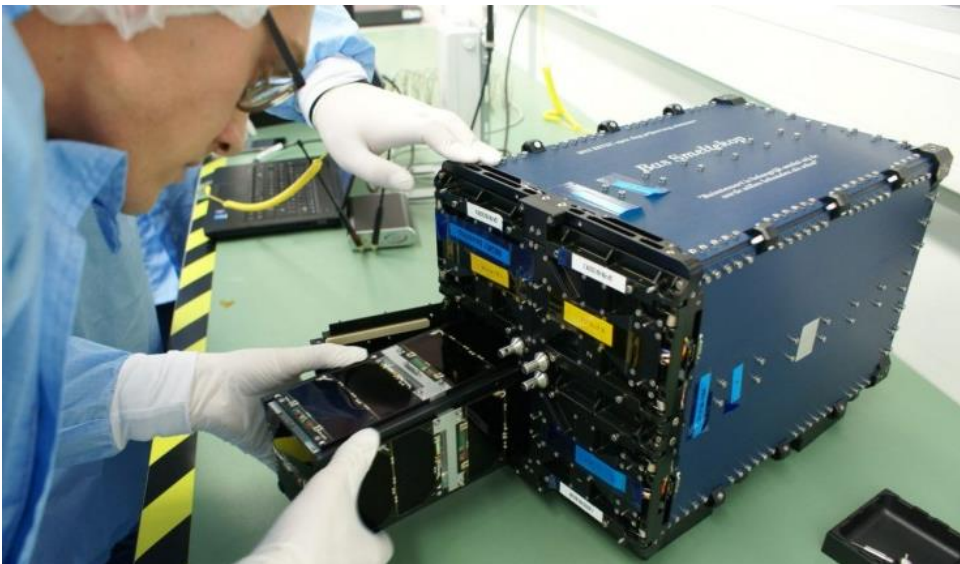


CubeLabs™

Standardized platform

Open architecture

Multiple biomedical applications
to run simultaneously and
independently



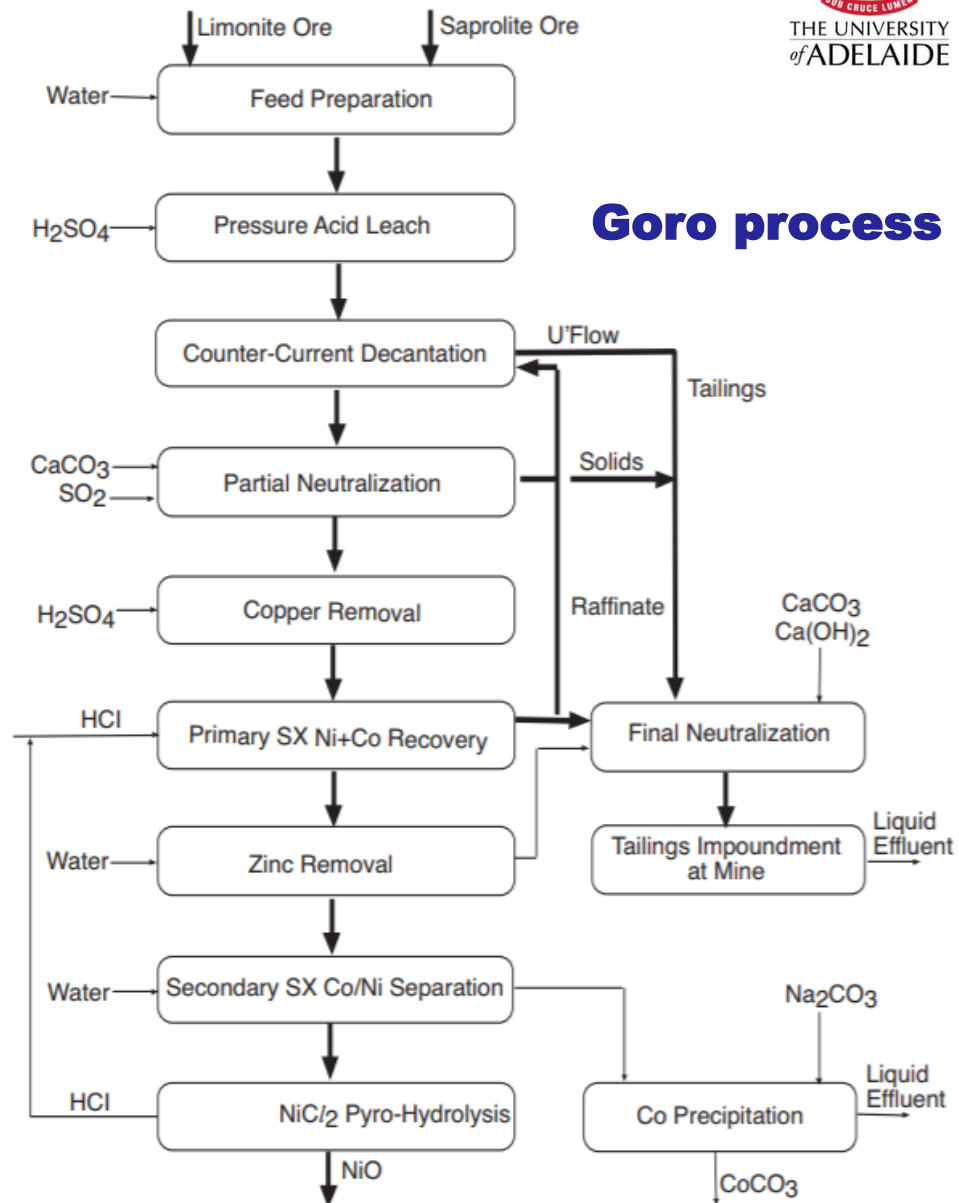
DIDO-2

10x10x30 cm

4.5 kg

Medical experiments

SOLVENT EXTRACTION IN MINING INDUSTRY



GLENCORE'S MUTANDA MINE - Largest Cobalt Reserve on Earth



S SMALL CAPS

HOT TOPICS LATEST TOP STOCKS SECTORS IPOs TOOLS

Glencore's Mutanda mine shutdown could prompt earlier than expected cobalt price revival

By *Lorna Nicholas* - August 14, 2019



PLANET VS ASTEROID MINING

MOON *or* ASTEROID

NEAR CIRCULAR EARTH ORBIT
Average Distance 240K Miles

FEASIBLE GRAVITY WELL
One-Sixth Earth's Gravity

VALUABLE NATURAL RESOURCES
Helium-3, Water, Silicon, Rare Metals

DEEP SPACE MISSION STAGING
Fuel, Supplies, Gear, Training

H₂O EXTRACTABLE AT POLES
Propellant, Potable Water, Air

VERY ACCESSIBLE ORBITS
Intermittent Near-Earth Passes

LESS FUEL & LOWER THRUST
For Return to Earth Orbit

RICH RANGE OF RESOURCES
Water, Methane, Rare Metals

PERPETUAL SUNLIGHT
Continuous Solar Electric Power

H₂O READILY EXTRACTABLE
Minimal Gravity, Less Cohesion



ASTEROIDS AND THEIR VALUE MATERIALS


Table 1. Metal contents and distributions in the diverse types of most valuable asteroids

| Name | Type | Discovered year | Composition | MOID (AU) | Value (\$) | Δv (km/s) |
|------------|------|-----------------|---|-----------|----------------|-------------------|
| Ryugu | Cg | 1999 | Ni, Fe, Co, H ₂ O, N ₂ , H ₂ , NH ₃ | 0.000638 | 82.76 billion | 4.663 |
| 1989 ML | X | 1989 | Ni, Fe, Co | 0.082029 | 13.94 billion | 4.889 |
| Nereus | Xe | 1982 | Ni, Fe, Co | 0.003153 | 4.71 billion | 4.987 |
| Bennu | B | 1999 | Fe, H ₂ , NH ₃ , N ₂ | 0.003223 | 669.96 million | 5.096 |
| Didymos | Xk | 1996 | Ni, Fe, Co | 0.039777 | 62.25 billion | 5.162 |
| 2011 UW158 | Xc | 2011 | Ni, Fe, Co, Pt | 0.002914 | 6.69 billion | 5.189 |
| Anteros | L | 1943 | MgSiO ₃ , Al, Fe, Fe ₂ O ₄ Si | 0.062212 | 5.57 trillion | 5.44 |
| 2001 CC21 | L | 2001 | MgSiO ₃ , Al, Fe ₂ O ₄ Si | 0.083067 | 147.04 billion | 5.636 |
| 1992 TC | X | 1992 | Ni, Fe, Co | 0.167212 | 84.01 billion | 5.648 |
| 2001 SG10 | X | 2001 | Ni, Fe, Co | 0.017183 | 3.05 billion | 5.88 |
| 2002 DO3 | X | 2002 | Ni, Fe, Co | 0.029415 | 334.44 million | 5.896 |
| 2000 CE59 | L | 2000 | MgSiO ₃ , Al, Fe ₂ O ₄ Si | 0.008298 | 10.65 billion | 6.015 |
| 1995 BC2 | X | 1995 | Ni, Fe, Co | 0.135685 | 78.87 billion | 6.01 |
| 1991 DB | C | 1991 | Ni, Fe, Co, H ₂ O, N ₂ , H ₂ , NH ₃ | 0.102803 | 168.20 billion | 6.146 |
| 2000 RW37 | C | 2000 | Ni, Fe, Co, H ₂ O, N ₂ , H ₂ , NH ₃ | 0.008221 | 29.27 billion | 6.225 |
| 1998 UT18 | C | 1998 | Ni, Fe, Co, H ₂ O, N ₂ , H ₂ , NH ₃ | 0.037188 | 644.70 billion | 6.221 |
| Seleucus | K | 1982 | Ni, Fe, Co, H ₂ O, N ₂ , H ₂ | 0.102357 | 33.52 trillion | 6.289 |

Asterank
scientific and
economic
database of
over 600'000
asteroids

WATER FOR ASTEROID MINING: NEED FOR DISRUPTIVE TECHNOLOGIES TO GIVE ECONOMIC CASE



- ❑ South Australia's cobalt ~2900 t/a in Mount Gunson Copper Mine
- ❑ Largest mine on Earth: Glencore's Mutanda, Republic of Congo, 23,900 t cobalt in 2017
- ❑ 190 tons of water per ton of cobalt (190 : 1)
- ❑ 1 ton water from Earth to LEO: 3 million \$  80 kdollar per t cobalt on Earth
- ❑ Ryugu (1999 JU₃): 4.5 x 10⁶ t cobalt
- ❑ 80 kdollar per t cobalt, Earth <-> 300k per t water, Space 2025

METAL CONCENTRATION 10 mol/l (space) vs. 0.1 mol/l (Earth);

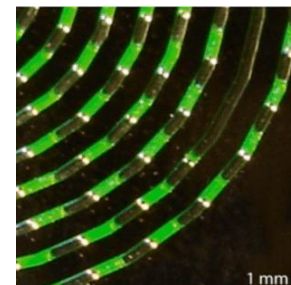
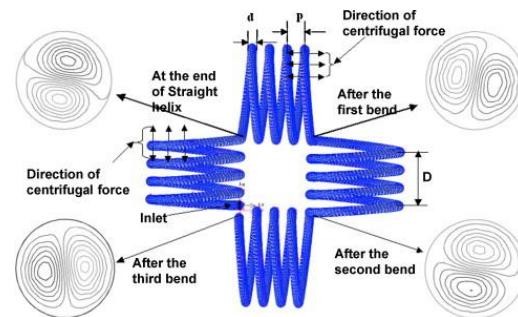
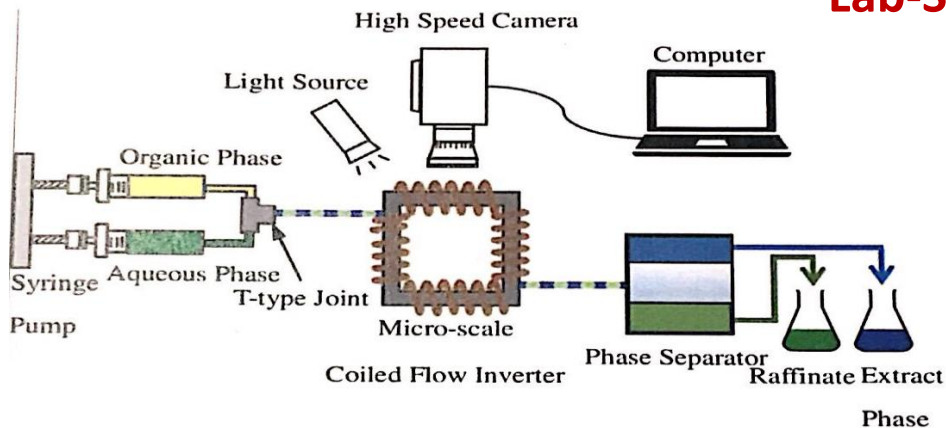
RELATIVE METAL LOAD i.e. Ni:Co = 10:1 in space vs. Ni:Co = 3:1 on Earth;

METAL MIXTURE COMPLEXITY i.e. Fe, Ni, Co, Pt, ... ;

ALLOYS i.e. Kamacite in space, α -(Fe, Ni); Fe_{0.9}Ni_{0.1}

SEGMENTED FLOW IN A COILED FLOW INVERTER

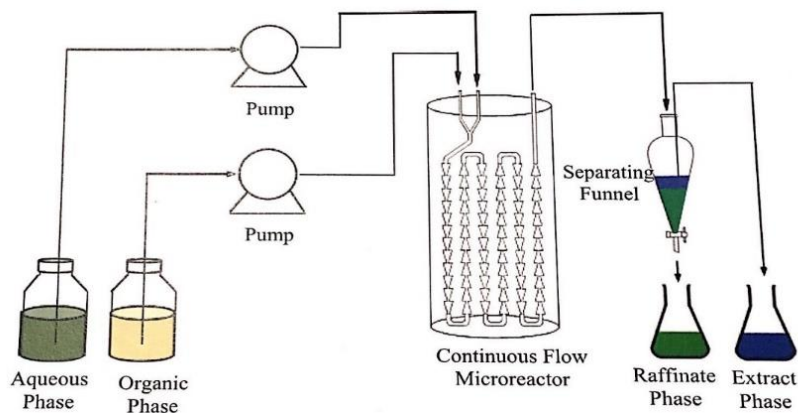
Lab-Scale: Coiled-Flow Inverter (8 l/h)



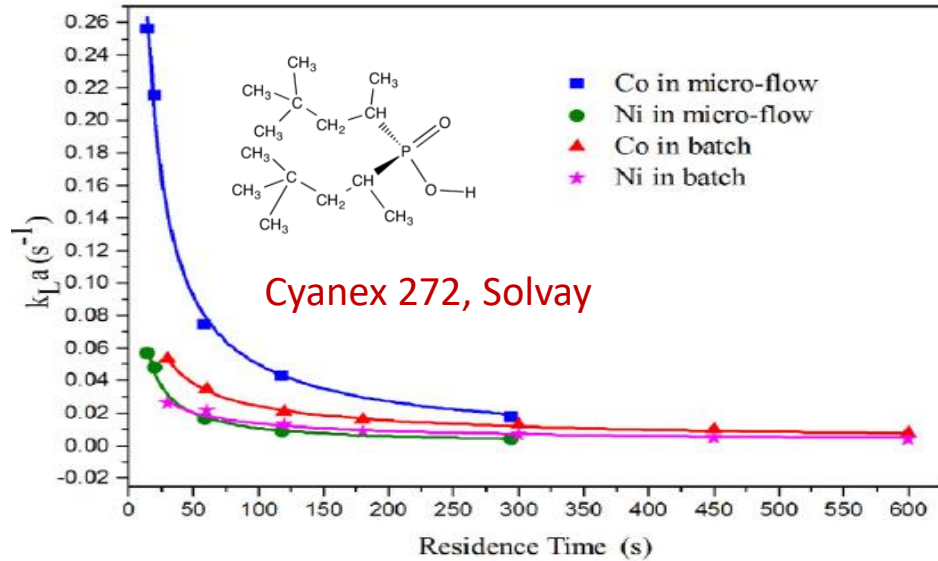
Production-Scale: Coiled-Flow Inverter (1700 l/h)



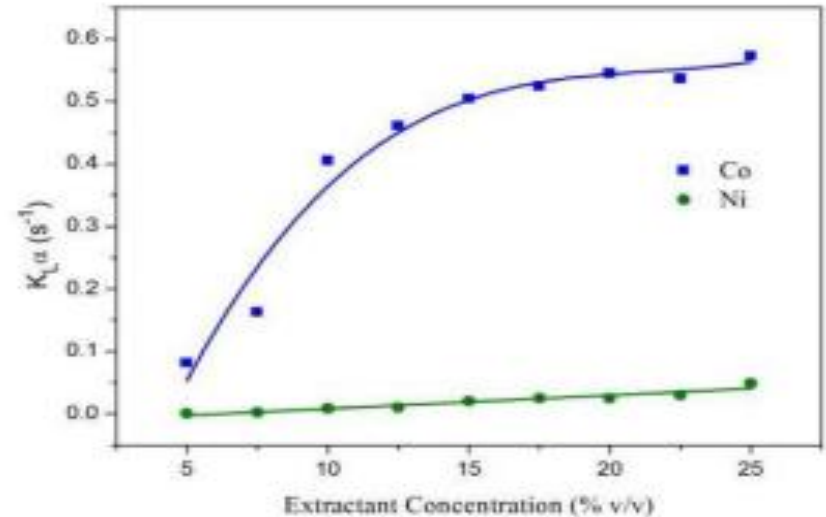
Pilot-Scale: Re-entrance Flow Reactor (360 l/h)



COBALT VS NICKEL ENRICHMENT BY CONTINUOUS-FLOW



Volumetric mass transfer coefficient k_{La} in coiled micro-flow inverter



k_{La} in re-entrance flow reactor

- Continuous-flow extraction 7 times more selective, and 10 times faster than batch
- Chance for substantial water savings and much higher productivity at small footprint

DISRUPTIVE FLUID PROPERTIES

- mimicking 'space fluids'



Nickel nitrate
10 mol/l

'wet liquid metal'

182 g metal salt

< 100 ml water

Almost 3 kg/l

4 x water viscosity

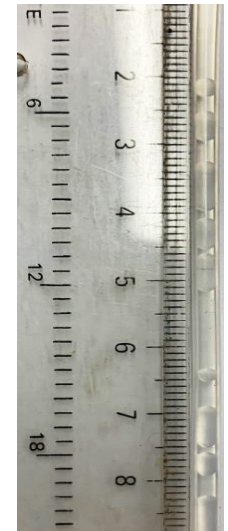
Pure nickel nitrate –
melting point 56 C –
solvent-free process?

FLOW PATTERNS OF WATER-KEROSENE WITH DIFFERENT MICRO-FLOW CONTACTORS

Y-junction
60 mL/h; 1:1

T-junction 180°
60 mL/h; 1:1

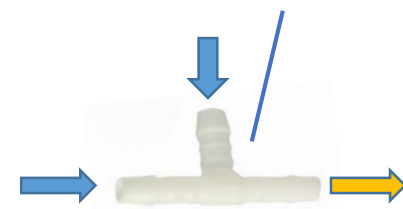
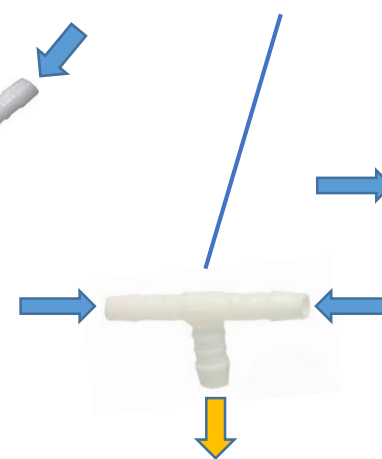
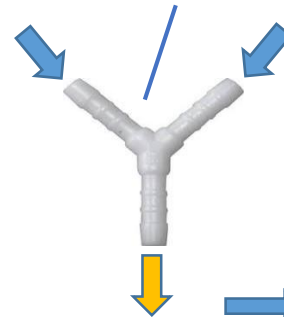
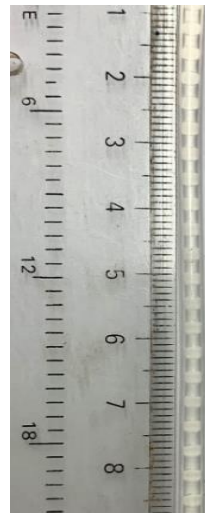
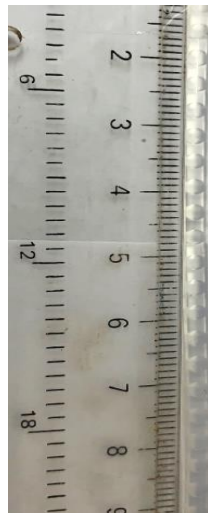
T-junction 90°
60 mL/h; 1:1



Y-junction
720 mL/h; 1:1

T-junction 180°
720 mL/h; 1:1

T-junction 90°
720 mL/h; 1:1



FLOW PATTERNS OF WATER-KEROSENE VS METAL LOAD AND FLOW RATE

Y-junction
240 mL/h; 1:1
all 3 cases

Y-junction
480 mL/h; 1:1
Zero Ni,Co

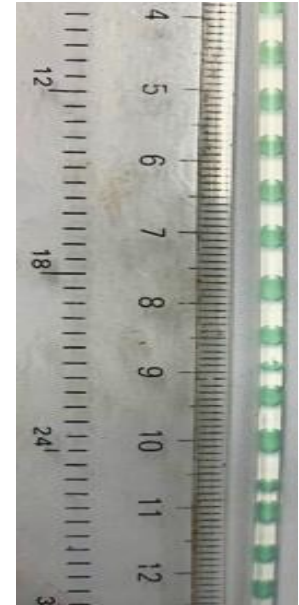
Y-junction
480 mL/h; 1:1
Ni: 0.5 mol/l
Co: 0.05 mol/l

Y-junction
480 mL/h; 1:1
Ni: 10 mol/l
Co: 1 mol/l

Y-junction
720 mL/h; 1:1
Ni: 10 mol/l
Co: 1 mol/l

In no case
regular flow
pattern

Irregular slugs



IONIC LIQUIDS



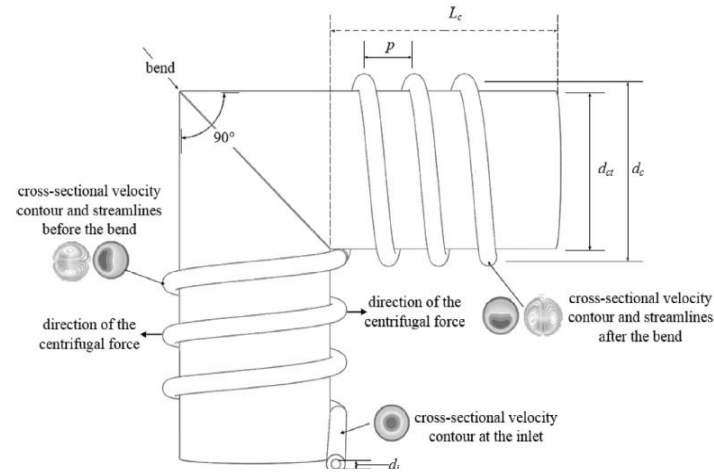
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| Ionic liquid | Metal ion | Mechanism | Ligand/solvent | Performance (E %) | Stripping agent | reference |
|--|---|---|--|---|--|-----------|
| [HMIM][BF ₄] | Co(II) | IE ^a | NaCl | Co(II), D= 5.8 | NaPF ₆ (0.03 M): Co(II),100% | [34] |
| [HJMT][Cy272]/ Kerosene, Exxon D100 and Solvesso 200 | Co(II), Ni(II) | IE | HCl (1 M) | Co(II),>99%; Ni(II),11% | EDTA (0.02 M): Co(II) ,95%; Ni(II), 83% | [35] |
| [BuGBOEt][Dca] | Cu(II), Ni(II), Pb(II), Cd(II) | | | Pb(II), 858%;Cd(II), 95%;Ni(II), 82%;Cu(II), 83% | EDTA (0.1 M): Ni(II), 100%; Cu(II), 100% | [36] |
| [BuGBOEt][Tf ₂ N] | Cu(II), Ni(II), Pb(II), Cd(II) | | | Pb(II), 38%; Cd(II), 41%; Ni(II), 20%; Cu(II), 22% | EDTA (0.1 M): Ni, 100%; Cu, 100% | [36] |
| [Dibutyl IM][Br] | Co(II), Ni(II) | AE ^b | Acidic thiocyanate HNCS | Co(II), >99% | NH ₃ (1.0 M): Co(II), >93% | [37] |
| [N ₈₈₈₈][oleate] | Zn(II), Co(II) ,Ni(II) | IL metal complex | HCl | Co(II), D=20.4; Ni(II), D=13.5; Zn(II) D>200 | | [38] |
| BuNC ₂ OC ₄ -Sac, | Cu(II), Ni(II), Co.(II), Pb(II), Cd(II) | cation exchange and IP ^c | | Cu(II), 30%; Ni(II), 5%; Co.(II), 10%; Pb(II), 20%; Cd(II), 95% | EDTA (0.1 M) | [39] |
| BuNC ₂ OC ₄ -Clisal | Cu(II), Ni(II), Co.(II), Pb(II), Cd(II) | cation exchange and IP | | Cu(II), 100%; Ni(II), 96%; Co.(II), 94%; Pb(II), 100%; Cd(II), 100% | EDTA (0.1 M): Ni(II), 100%; Co(II), 100%, Pb(II), 100%; Cd(II), 100% | [39] |
| BuNC ₂ OC ₄ -Dca | Cu(II), Ni(II), Co.(II), Pb(II), Cd(II) | cation exchange and IP | | Cu(II), 98%; Ni(II) 98%; Co.(II), 95%; Cd(II), 95% | EDTA (0.1 M): Ni(II), 100%; Co(II), 100%; Pb(II), 100%; Cd(II), 100% | [39] |
| Cyphos IL 101 | Co(II), Ni(II) | AE and split- anion | H ₂ SO ₄ , (NH ₄) ₂ SO ₄ | Co(II),>90% | Water: Co(II)>99% | [40] |
| [P ₄₄₄₁₄][Cl] | Co(II), Ni(II) | - | NaCl | Co(II), D=100; Ni(II), D=0.2 | - | [41] |
| [A336][CA-12] | Co(II), Ni(II) | ion association | Toluene, Na ₂ SO ₄ | Co(II), D>95% ; Ni(II), D>65% | H ₂ SO ₄ : Co(II), 100%; Ni(II)>99% | [41] |
| [A336] ₂ SO ₄ | | | | | | |
| [P ₆₆₆₁₄][Cl] | Cu(II), Co.(II), | - | HCl (9 M) | Co.(II), > 98%; Fe(III), >99% | (EDTA + Water): Fe(II)>80% | [42] |

Ionic liquids are powerful in metal extraction: faster & more selective; can potentially be operated openly

DEAN NUMBER WATER

- Curvature & Miniaturisation

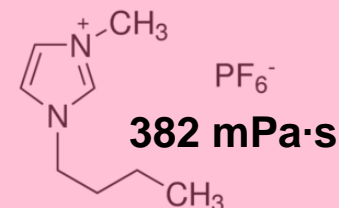
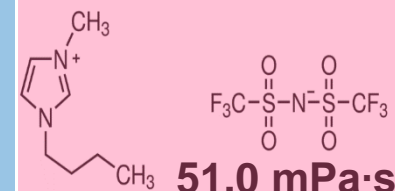


| Flow rate Q (ml/min) | U (m/s) | Re | Ca | We | d_i (mm) | d_c/d_i | Dn |
|------------------------|-----------|--------|--------|--------|------------|-----------|--------|
| 16 | 0.3392 | 371.6 | 0.0057 | 2.1381 | 1 | 5 | 166.19 |
| 18 | 0.3816 | 418.08 | 0.0064 | 2.7061 | 1 | 5 | 186.97 |
| 20 | 0.424 | 464.53 | 0.0071 | 3.3408 | 1 | 5 | 207.74 |
| 16 | 0.3392 | 371.6 | 0.0057 | 2.1381 | 1 | 10 | 117.52 |
| 18 | 0.3816 | 418.08 | 0.0064 | 2.7061 | 1 | 10 | 132.21 |
| 20 | 0.424 | 464.53 | 0.0071 | 3.3408 | 1 | 10 | 146.90 |
| 16 | 0.0331 | 294.47 | 0.0006 | 0.1655 | 3.2 | 20 | 25.968 |
| 18 | 0.0372 | 331.27 | 0.0006 | 0.2094 | 3.2 | 20 | 29.214 |
| 20 | 0.0414 | 368.08 | 0.0007 | 0.2585 | 3.2 | 20 | 32.460 |

DEAN NUMBER IONIC LIQUID

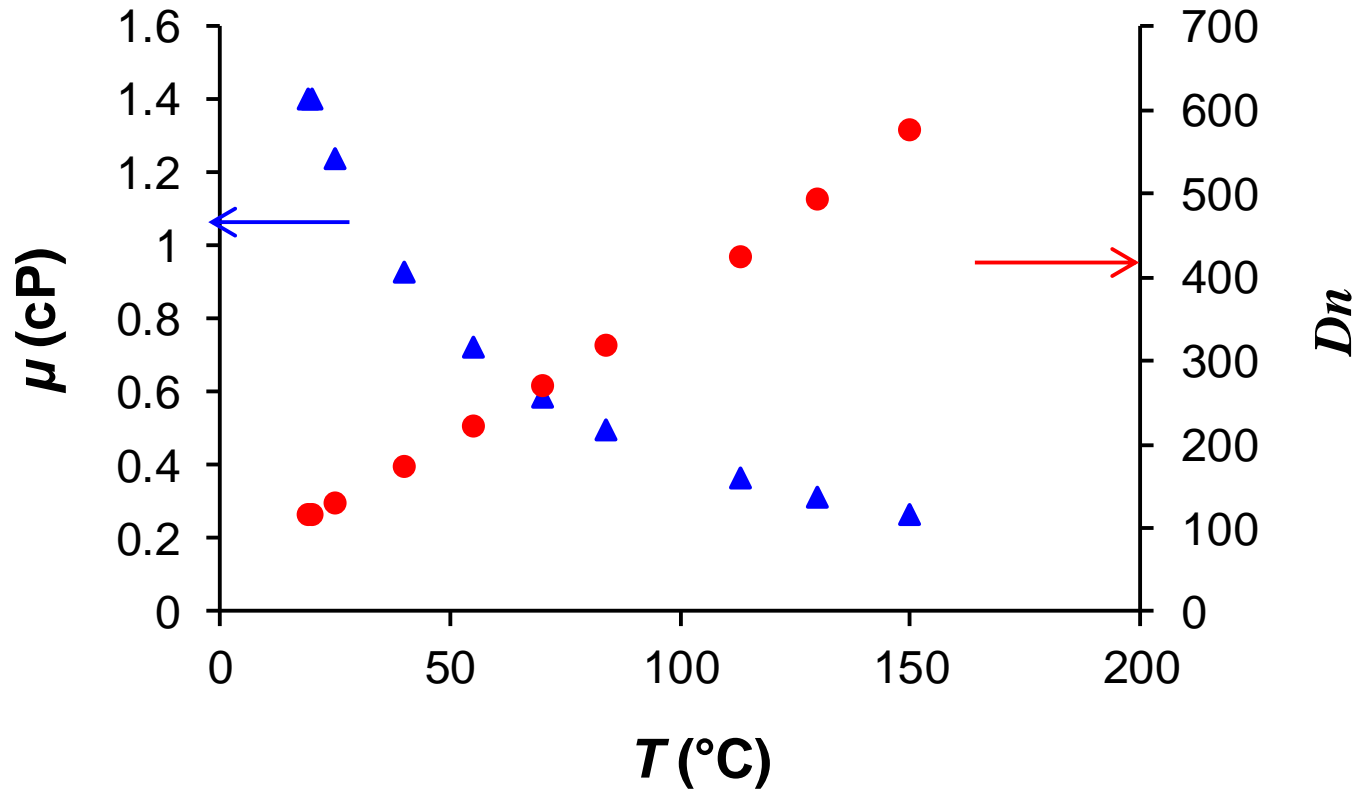
| Flow rate Q (ml/min) | U (m/s) | Re | Ca | We | d_i (mm) | d_c/d_i | Dn |
|---------------------------|-----------|--------|--------|--------|------------|-----------|--------|
| 16 | 0.3392 | 371.6 | 0.0057 | 2.1381 | 1 | 10 | 117.52 |
| 18 | 0.3816 | 418.08 | 0.0064 | 2.7061 | 1 | 10 | 132.21 |
| 20 | 0.424 | 464.53 | 0.0071 | 3.3408 | 1 | 10 | 146.90 |
| 16 | 0.3392 | 9.71 | 1.336 | 12.979 | 1 | 10 | 3.07 |
| 18 | 0.3816 | 10.93 | 1.503 | 16.426 | 1 | 10 | 3.46 |
| 20 | 0.4240 | 12.14 | 1.670 | 20.279 | 1 | 10 | 3.84 |
| 16 | 0.3392 | 2.25 | 6.9290 | 15.623 | 1 | 10 | 0.713 |
| 18 | 0.3816 | 2.54 | 7.7951 | 19.773 | 1 | 10 | 0.802 |
| 20 | 0.4240 | 2.82 | 8.6612 | 24.411 | 1 | 10 | 0.891 |

WATER
0.89 mPa·s



DEAN NUMBER & VISCOSITY

NaCl (aq) - Brine



Salt solutions at extremely high concentrations (molar) will fluidically benefit from high-T through viscosity reduction

ORBITAL ECONOMICS

Easily Recoverable Asteroids

Rocket equation

$$\frac{m_f}{m_i} = \exp\left(\frac{-\Delta v}{v_{ex}}\right)$$

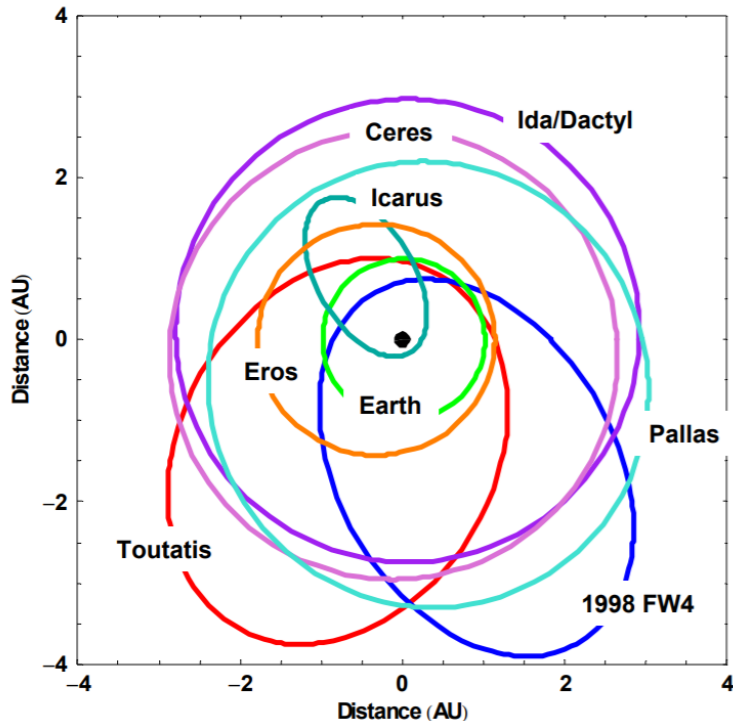
$m_f \equiv$ final mass

$m_i \equiv$ initial mass

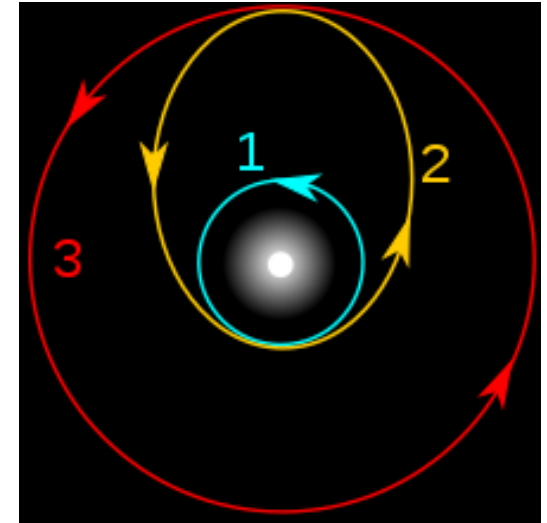
$\Delta v \equiv$ velocity increment

$v_{ex} \equiv$ exhaust velocity

- Selected asteroid trajectories



Hohmann
transfer: delta-v



| Mission | Δv |
|--------------------------------------|-----------------------------------|
| Earth surface to LEO | 8.0 km/s |
| LEO to near-Earth Asteroid | 5.5 km/s [note 1] |
| LEO to lunar surface | 6.3 km/s |
| LEO to moons of Mars | 8.0 km/s |

ASTEROID COMMUNICATION

Communication Challenges

- ground support for the mission,
- trajectory corrections,
- remote control in case of emergencies,
- transmission of data for analysis



Usada Deep Space Center

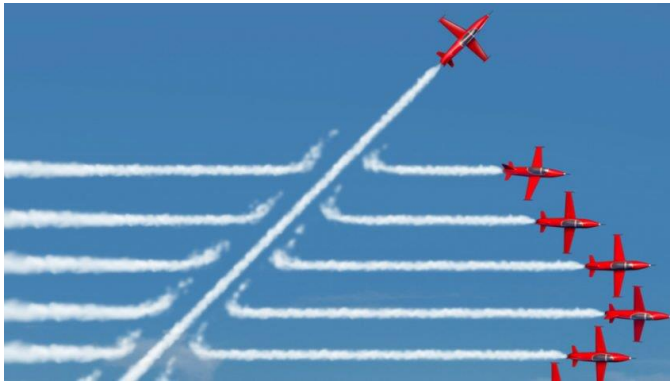
Technology High and low gain antennas, Deep Space Network (DSN), X-band communication, transponder, amplifier, ...

Expenditure Typical communication module: 10 kg weight, 120 W power (10% of typical power consumption in an aircraft)

Limitations Low data rate, high latency, delay time in response, only 12 antennas in DSN

CONCLUSIONS

- ❑ Space mining is more than automation of remote processes
- ❑ Disruptive technologies are not considered in current ISRU approaches, yet are the key to economics



- ❑ Holistic picture needed – interdisciplinary
- ❑ The economics themselves have to be developed = business case
- ❑ Need for a business case – space mining trade
- ❑ Off-earth thinking for Earth?

SPACE HORROR – SPACE POLLUTION





AFCS 19



Australian Flow Chemistry Symposium

2-3 December 2019
Melbourne, Australia



International Conference of Microreaction Technology

8-11 December 2020
Melbourne, Australia

